HOUSEHOLD CHEMISTRY FOR GIRLS: A LABORATORY GUIDE

Published @ 2017 Trieste Publishing Pty Ltd

ISBN 9780649441679

Household Chemistry for Girls: A Laboratory Guide by Jamie Maud Blanchard

Except for use in any review, the reproduction or utilisation of this work in whole or in part in any form by any electronic, mechanical or other means, now known or hereafter invented, including xerography, photocopying and recording, or in any information storage or retrieval system, is forbidden without the permission of the publisher, Trieste Publishing Pty Ltd, PO Box 1576 Collingwood, Victoria 3066 Australia.

All rights reserved.

Edited by Trieste Publishing Pty Ltd. Cover @ 2017

This book is sold subject to the condition that it shall not, by way of trade or otherwise, be lent, re-sold, hired out, or otherwise circulated without the publisher's prior consent in any form or binding or cover other than that in which it is published and without a similar condition including this condition being imposed on the subsequent purchaser.

www.triestepublishing.com

JAMIE MAUD BLANCHARD

HOUSEHOLD CHEMISTRY FOR GIRLS: A LABORATORY GUIDE

Trieste

HOUSEHOLD CHEMISTRY FOR GIRLS

8

A LABORATORY GUIDE

BY

JAMIE MAUD BLANCHARD

HEAD OF THE CHEMISTRY DEPARTMENT IN THE LOS ANGELES HIGH SCHOOL, CALIFORNIA

0.000

2

ľ

Boston ALLYN AND BACON 1912

200

.

COPYRIGHT, 1912, BY JAMIE MAUD BLANCHARD.

12

((*))

 (2π)

i a

!

ñ

43

81. 8

.3

23

ł

Norizood Bress J. S. Cushing Co. - Berwick & Smith Co. Norwood, Mass., U.S.A.

PREFACE.

To outline a strong course in chemistry, especially suited to girls of high school age, is the purpose of this book. Several years' experience in developing this course has strengthened the author's conviction that it is better in every way to give girls special work in chemistry rather than to make their study identical with that for boys. In the Los Angeles High School, where this plan has been followed for some time, it has brought about a gratifying increase in the number of girls studying chemistry.

But it was with no intention of weakening the course or of presenting to girls merely the "pyrotechnics" of chemistry that the differentiation has been made. On the contrary, the same standard of thoroughness has been maintained for both boys and girls, with the result that full credit has been granted by the authorities of the University of California for the year's work in homehold chemistry.

Though our ultimate aim is the training of intelligent homemakers, this is a manual of chemistry, not of domestic science. It is therefore suitable for a purely academic high school, no less than for a polytechnic high school, where a rigorous course in household chemistry forms a necessary foundation for the work in domestic science. The choice of subjects is based in a general way on the following scheme:

What we breathe.

What we drink and use for cleansing.

What we use for fuels and illuminants.

Chemical nature of common substances.

iii

PREFACE.

Foods and food values.

Adulterants and simple methods for their detection.

Textiles - care of textiles, removal of stains, etc.

The principles of inorganic chemistry — to which the first half of the book is devoted — are developed from a study of common substances, with special reference to their life interest and practical applications. This offers a scientific treatment of the subject, while making it more useful and attractive to the student.

The second half of the book, beginning with Experiment 28, is devoted to qualitative experiments in organic chemistry, as delicate quantitative experimentation is beyond the ability of high school pupils. Supplementary reading is of course advisable in this connection; with this in view, a full list of library text-books is given, and definite references to these accompany the experiments.

For the convenience of teachers, an alphabetical list of the necessary apparatus and chemicals is included, though the laboratory equipment differs but slightly from that of the regular course for boys. Other features, such as a list of exhibits and charts, suggestions for work at home, references for further reading along lines that are not so essential, but are of allied interest, have also been added.

Los ANGELES, January, 1912.

iv

CONTENTS.

n	reation	s for Laboratory V	Vork								PAG
	recuor	is for Deboratory	TOLK		•2" 5780 5	•2 1995		- - -		•	3
400		Part I.	Ino	rganic	Che	emis	try.				
1.		luction.									
		Physical Change		3.		•	٠	12	•	•	1
		Chemical Change	10		6 2	•	藻	23		.3	
	э.	Study of a Match	28	2	1	25	10	25	9 9 2	8 8	1
2.	The Atmosphere.										
		Study of Oxygen		3.03	9 3	*	3 1	3¥			1
		Study of Nitrogen		(1 .	20			10	3. • 3	.	1
		Study of Carbon I					. And				1
		Separation of the									1
	8.	Presence of Carbo	n Dio	zide an	d W	ator	Vapo	r in t	the A	ir.	1
	· 9.	Ventilation .	35	((2))	20	1 20	22	(1 ²³	1.000	20	2
8.	Water.										
	10.	Study of Hydroge	α.	0.00	•3		242	200		-	2
	11.	Study of Water -	Elect	rolysis		18. I.					2
	12.	Study of Water -	Decor	npositi	on b	y Ch	lorin	в.		2	2
	13.	Study of Water -	Synth	esis	4 3	S.,		1.94	200	20	2
	14.	General Distributi	ion of	Water				1.4			2
	15.	Simple Tests for Impurities in Water									
	16,	Distillation - A M	leans	of Puri	fyin	g Wa	ater	3. 2	(1 4)	3 2	8
	17.	To show the Solve	ent Po	wer of	Wat	ег	*	38	1.0	52	3
é.	Fuels	and Illuminants.									
_	18.	Carbon-distributio	m.	-	2.1	~		5 8		÷0	3
	19.	Study of Amorph	ous Ca	rbon				100			3
	20.	The Union of Car			gen		3		199		3
	21.				24		32		0.00	20	8
	. 22.	Study of Flames		0.00			÷				3
	23.	Common Gaseous	Hydro		18		18	8			4
	24.	Study of Some Li	quid H	ydroca	rbo	ns.				2	4
δ.	The Nature of Common Substances.										
		A Study of Acida			12	3	32	-	11	80	4
		A Study of Bases		37275			-	14	1910		4
	27.	A Study of Salts		20000		÷.					4
		re round or come	1.5			201			1.0	5 2	20

.

•

.

CONTENTS.

Part II. Simple Organic Chemistry.

0.	Foods.											PAGE	
	28.	8. To determine the Amount of Water in Foods											50
	29.	Starch						10					51
	30.	Sugara	•0	\sim		196	2003	÷	10		138		58
	81.	Digestion	of 8	tarch				•					55
	32.	Cellulose	•		¥.	2				ě		1	56
	33.	Gums, Re	sins,	Vari	ish	es .		•			12		57
	34.	Nitrogeno	us S	ubstan	aces			• 22					58
	35.	Special Pr	otei	ls .							1		60
	36.	Special Ge	elatin	noids	2		1.0	10		÷.	21		61
	37. Digestion of Nitrogenous Substances											63	
	38.	Fats and						•	•		~~	- 100	65
	39.	Mineral M	latte	r in F	ood	s .	5	7755		8	行		67
	40.	Milk .	2 0	340	22	82	-	20	44			-	68
	41. Leavening Agenta: Baking Soda, Baking Powders												70
		Practical									1000	1	72
		Yeast - A				· · · · · ·		÷ 8.	- 33	- 2	22	(5	78
		Practical					1.		-	~	37	27	75
		Tea Col								- C		225	77
	48.	Food Valu	aes.	Men	115	2			10		1	1	79
	Food Adulterations.												
•												80	
	 Tests for Adulterants in Milk 48. Tests for Adulterants in Butter, Meat Producta, Mustard, 											00	
	40,	Baking			QS 11	n isung			1041	ICER, D	AIDSU	aru,	81
	40	Tests for		211.07.00	1			Talli		Dartus.			01
	49.	cream		ntera.	nus	m Ca	may,	Jem	es, I	CXLFav	sus,	Icc-	83
		cream	U.)	2.2	- 22		1		10	•			09
8.	Soaps.												
	50.	Soaps - I	Iard	Wate	r.	\sim		3.023	12	302	22		85
9.	Clothing.												
		Textiles -	- Fib	108					£2.		121		86
		Stains			1	- <u>S</u>		•	23		•	10	88
		Comus	1960	83	8	25	88	3120	70	12	0	22	(783)
A	pendis	ε.											
	Ref	erences for	Clas	s Stu	dy	•						12	91
	List	t of Exhibit	ta							1	<u>.</u>	84	96
	Bib	liography		÷1				2002	*2			()•	98
	List	t of Supplie	28										100
		7 G CARLON (1997) (1993)											
								•					

vi

.

HOUSEHOLD CHEMISTRY FOR GIRLS.

DIRECTIONS FOR LABORATORY WORK.

The Table. — When the work is completed, all apparatus should be removed from the table and the top carefully cleaned. The drawer should be lined with clean paper. Apparatus easily broken, as test-tubes and beaker, should be kept in boxes.

The Burner. — To light the Bunsen burner, turn the gas on full, and apply a lighted match about four inches above the burner; then lower the gas to the desired height. If the match is held close to the burner, the flame is liable to "strike back," that is, to burn at the base. In such a case, turn off the gas, let the burner cool, and then relight it. It is dangerous to let the base of the burner become heated, as the connecting rubber tubing melts and allows a large stream of gas to burn. The flame should be a clear blue and deposit no soot.

The Apparatus. -(a) The names of the pieces of apparatus should be learned at the outset.

(b) All apparatus should be clean before work is begun. Clogged connecting-tubes will cause an explosion.

(c) In using a ring-stand, place it at your right hand and extend the apparatus to the left.