EXPERIMENTAL CHEMISTRY FOR JUNIOR STUDENTS. PART I. INTRODUCTORY

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Experimental Chemistry for Junior Students. Part I. Introductory by J. Emerson Reynolds

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J. EMERSON REYNOLDS

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EXPERIMENTAL CHEMISTRY.

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EXPERIMENTAL CHEMISTRY

FOR

JUNIOR STUDENTS.

BY

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PART I.

INTRODUCTORY.

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PREFACE.

This work is identical in plan with my Six Lecture's on Experimental Chemistry, but different in style, and much extended in range, so as to include the amount of knowledge of fact and principle usually expected from junior Arts, Medical and Pharmaceutical Students, as well as from the higher classes in Intermediate Schools.

The system pursued in this book is designed to lead the student through a series of connected experiments—often quantitative in form—and to assist him in the interpretation of his results, and in devising modes of testing the validity of the conclusions drawn from them. Thus while acquiring a tolerably wide knowledge of fact, the student receives a certain amount of training in the purely 'experimental method' of investigating Nature. If this training be sound, the study of Chemistry must prove a valuable means of mental education. How far the particular plan pursued in the following pages is likely to contribute to such a result, I must leave others to judge:

but a reviewer of my Lectures was so good as to say :- 'In these Lectures the author departs widely from the usual routine of elementary treatises. . . . We believe that he is right in the plan he has adopted, and that instruction of this would greatly facilitate the acquisition of clear and distinct ideas of the leading facts and laws of Chemistry.' (Chemical News, vol. xxix. page 227.) This work is divided into four parts, each one being, as far as practicable, complete in itself. Part L is introductory, and deals with first principles, and with the chemistry of the typical elements, hydrogen and oxygen, and their compounds; Part II., with the rest of the non-metals; Part III., with the metals; and Part IV., with organic chemistry. The experiments described are, whenever possible, those easily performed; in some cases, however, methods are necessarily detailed which the student may not have either the skill or the means to carry out, but he should endeavour to see these operations carefully conducted. It is assumed throughout that the reader can obtain some practical instruction in glass working and the construction of apparatus.

It is only necessary to add that the complete work will contain the solutions of all the problems in my Lecture Note Book.

J. E. R.

CHEMICAL LABORATORY, TRINITY COLLEGE, DUBLIN: November, 1880.

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