THE PHYSICAL CHEMISTRY OF THE METALS

Published @ 2017 Trieste Publishing Pty Ltd

ISBN 9780649094332

The physical chemistry of the metals by Rudolf Schenck

Except for use in any review, the reproduction or utilisation of this work in whole or in part in any form by any electronic, mechanical or other means, now known or hereafter invented, including xerography, photocopying and recording, or in any information storage or retrieval system, is forbidden without the permission of the publisher, Trieste Publishing Pty Ltd, PO Box 1576 Collingwood, Victoria 3066 Australia.

All rights reserved.

Edited by Trieste Publishing Pty Ltd. Cover @ 2017

This book is sold subject to the condition that it shall not, by way of trade or otherwise, be lent, re-sold, hired out, or otherwise circulated without the publisher's prior consent in any form or binding or cover other than that in which it is published and without a similar condition including this condition being imposed on the subsequent purchaser.

www.triestepublishing.com

RUDOLF SCHENCK

THE PHYSICAL CHEMISTRY OF THE METALS



THE PHYSICAL CHEMISTRY OF THE METALS

BY

RUDOLPH SCHENCK

Professor of Physical Chemistry in the Technischen Hochschule in Aachen

TRANSLATED AND ANNOTATED BY

REGINALD SCOTT DEAN

Research Metallurgist, American Zinc, Lead and Smelting Co., St. Louis, Mo.

FIRST EDITION



NEW YORK

JOHN WILEY & SONS, INC.

LONDON: CHAPMAN & HALL, LIMITED

1919

PREFACE

There is perhaps no field where physical chemistry is of more value than in the field of metallurgy and metallography. In the latter field its usefulness has been recognized almost from the first but in the field of metallurgy its application has not been so general. It is hoped that this translation may aid in making the value of chemical dynamics and equilibrium clear to the metallurgist and metallurgical student.

Such additions as have seemed necessary have been incorporated in the text and the numerical data have been revised to agree with the accepted values. I have deemed it advisable not to go into the recent investigations concerning the electron theory since the scope of the work did not seem to warrant an extended treatise on this subject.

The book has been changed from lecture to text-book form and the references shifted from the appendix to the body of the book.

My thanks are due to Mr. A. T. McPherson of the U. S. Bureau of Standards for reading the manuscript as well as for many valuable suggestions. I wish also to thank Dr. Edward Schramm, director of this laboratory, for his encouragement and cooperation in the work of translation.

REGINALD S. DEAN.

Sr. Louis, Mo. March, 1919.



PREFACE TO THE GERMAN EDITION

This little book is the outcome of a series of lectures which I delivered in 1907, in the "Technischen Hochschule," at Aachen. Their purpose was to show the engineers of the Rhenish industrial district, before whom they were delivered, the use of chemical statics and to deepen their understanding of smelting operations and metallurgical processes.

I have endeavored, especially, to develop the principles of equilibrium clearly and so far as possible by the use of pertinent examples.

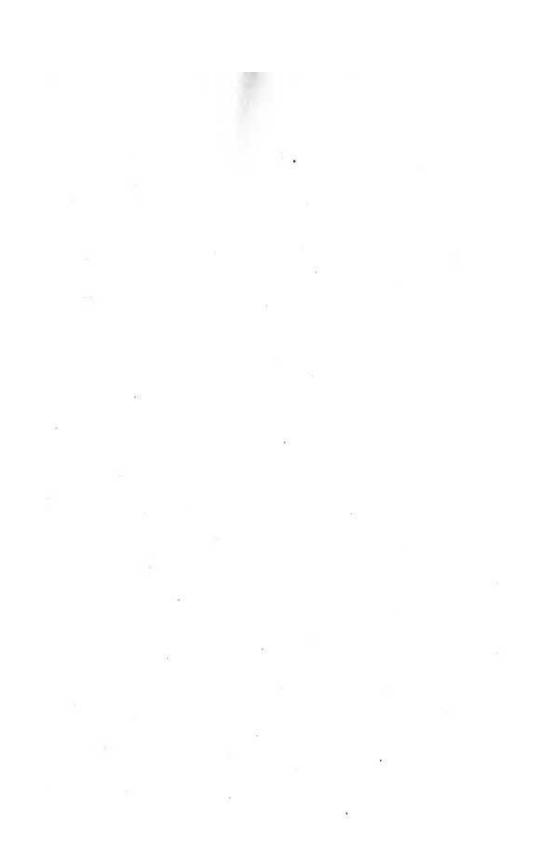
In the systematic survey of such a field new problems naturally arise and there is found in these lectures some heretofore unpublished data bearing on these problems, among which may be mentioned: the equilibrium between the various components of steel, the quantitative determination of amorphous carbon and graphite, and the investigation of the sulfatizing roast.

It has not been possible to consider here, all of the experimental matter relating to the physical chemistry of metals, but all fundamental questions have been treated rather thoroughly.

I wish to thank Dr. P. Goerens, for the preparation of the metallographs and photographs; Dr. Hemplemann, for his assistance in the preparation of the index, and Dr. Ratzbach, for the preparation of the diagrams.

THE AUTHOR.

AACHEN. July, 1908.



CONTENTS

D		PAGE
PREFACE TO GERMAN EDITION		
CHAPT	ER I	15
5 75 50		
INTRODUCTION, PROPERTIES OF METALS	******	1-38
Introduction, 1. Form Changes, 3. Monatomic State of Metal Vapors, 5. Pressure and the Definition of Melti Metals, 8. Density Change on Meltin Curves, 10. Polymorphy, 11. Ana to Melting, 14. Determination of the Antimony, 17. Enantiotropy and M 20. Electrical and Optical Properti Faraday's Law, 22. Metallic and Ga Theory, 23. Law of Wiedemann and Metals; Light Reflection and Absor Electron Concentration of Metals, Conductivity, 31. Thermo-Electric Fo	Color of Metal Vapors, 6. Vapor ing Point, 7. Melting Points of ag, 9. Heat of Fusion, 9. Cooling dogy of Polymorphic Transition Transition Point, 14. Explosive Ionotropy, 18. Crystal Growth, ies of Metals, Conductivity, 21. aseous Conduction, 22. Electron Franz, 23. Optical Properties of option, 27. Metallic Luster, 28. 30. Temperature Coefficient of	
CHAPTI	ER II	
METALLIC SOLUTIONS AND ALLOYS	***************************************	39-81
Colloidal Metal Solutions, 39. Dilut Pressure of Metallic Solutions, 42. Fre only Partially Miscible in the Liquid 47. Solidification Curve of Binary Zinc, 49. The Pattison Process, 52. S gation of Alloy Structure, 53. Solid Sc of Plating Platinum with Gold, 57. Metals Forming Solid Solutions, 57. Ternary and Quaternary Alloys, 64. Physical Properties to Alloy Structure sity of Alloys, 70. Potential of Allo Alloys, 73. Thermo-electric Force of	ezing Point of Alloys, 44. Metals State, 46. The Parkes Process, Alloys, the System Cadmium- iegregation in Alloys, 52. Investi- olutions, 55. The Heræus Process Crystallization Diagrams for Inter-metallic Compounds, 60. Relations of Mechanical and e, 68. Bearing Metals, 69. Den- oys, 72. Electrical Resistance of	

	PAGI
ALLOYS OF METALS WITH CARBIDES, OXIDES AND SULFIDES, IRON AND STEEL, MATTES, PHASE RULE	-138
Compounds with Metallic Properties, 82. The Iron—Carbon Alloys, 83. The Crystallization of the System Iron—Cementite, 85. Perlite and its Transition Products, 93. The Precipitation of Carbon in Iron—Carbon Alloys, 98. The Uses of Additions to Iron and Steel, 104. Alloys of Metals and Oxides; Copper—Copper Oxide, 110. Silver—Silver Oxide, 113. Alloys of Metals and Sulfides, 114. Lead—Lead Sulfide, 115. Antimony—Antimony Sulfide, 115. — Copper Copper Sulfide, 116. Silver—Silver Sulfide, 118. Iron—Iron Sulfide, 120. Nickel—Nickel Sulfide, 122. Alloys between Sulfides, 123. Phosphorus and Arsenic Containing Alloys, 125. Silicides of Metals, 131. The Phase Rule, 133.	
CHAPTER IV	
The Metallurgical Reactions, Oxideand Oxygen, 140. The Application of the Phase Rule, 140. Oxygen Tension of Oxides, 141. Oxygen Tension in Atmospheres, 143. The Equilibrium between two Oxides and Oxygen, 145. Direct Decomposition of Oxides by Heat, 148. Reduction by Metals, 149. Reduction by Gascous Reducing Agents, 149. Reduction by Hydrogen, 150. The Phase Rule, 150. LeChatelier's Principle, 151. The Mass Law, 152. Van't Hoff's Equation, 156. Reduction by Carbon Monoxide, 159.	-169
CHAPTER V	
DECOMPOSITION OF CARBON MONOXIDE, BLAST FURNACE PROCESS 170-	-204
Catalytic Decomposition of Carbon Monoxide, 170. Equilibrium between Carbon Monoxide and Iron, 173. The Analysis of Mixtures of Graphite and Amorphous Carbon, 178. The Blast Furnace Process, 180. The Mond Nickel Process, 202.	
. CHAPTER VI	
THE REACTIONS OF SULFIDES	-227